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# Elastic moduli of some rare-earth doped tellurite glasses

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ORIGINAL RESEARCH

#### Abstract:

Samarium oxide  $(Sm_2O_3)$  and ytterbium oxide  $(Yb_2O_3)$  doped tungsten-tellurite glasses (TWSm), (TWYb) in the forms of 80 (TeO<sub>2</sub>)-(20 - x)(WO<sub>3</sub>) - x(Sm<sub>2</sub>O<sub>3</sub>) and 80 (TeO<sub>2</sub>)-(20 - x)(WO<sub>3</sub>) - x(Yb<sub>2</sub>O<sub>3</sub>) with x = 0, 1.0, 2.0, 3.0, 4.0 and 5.0 mol% were prepared by the melt-quenching technique. The XRD analysis for the two prepared glass systems verified the samples' amorphous nature. The glass density ( $\rho$ ) was measured by the Archimedes method. Both ( $\rho$ ) and the molar volume ( $V_M$ ) of the two prepared systems showed an increase with increasing the concentrations of dopants Sm<sup>3+</sup> and Yb<sup>3+</sup>. Longitudinal  $V_L$  and shear  $V_S$  wave velocities were measured at 5 MHz frequency using the pulse-echo technique. Longitudinal (L), shear (G), bulk (K), and Young's moduli (E) have been founded. Moreover, Poisson's ( $\sigma$ ), Debye temperature ( $\theta_D$ ), softening temperature ( $T_S$ ), micro-hardness (H) have been founded for every glass sample. Elastic moduli showed an increase with increasing dopants concentration of (Sm<sub>2</sub>O<sub>3</sub>) or (Yb<sub>2</sub>O<sub>3</sub>). Moreover, the bulk modulus of (TWSm) glasses was higher than (TWYb) glasses. Mechanical properties of (TWSm) and (TWYb) are crucial parameter in manufacturing optical fibers.

Keywords: Glasses; Tellurite; Rare earth; Elastic moduli

# 1. Introduction

Tellurite glasses have been considered smart materials due to their unique physical properties and will be a candidate for sophisticated applications in comparison with other glasses [1–14]. Doping with rare earth (R.E.) or transition metal (T.M.) oxides induced changes in the glass structure and produced new physical properties for laser-fiber amplifier applications, color-changing in germano-tellurite niobate glass, 2–3  $\mu$ m mid-infrared luminescence, and Nd<sup>3+</sup>doped for transparent tellurite ceramics bulk lasers [15–19]. While in the year 2022, many researchers have been attracted to tellurite glasses in different ways like; spectroscopic measurements for solid-state lighting applications, fabrication, laser materials, magneto-optic fiber current sensors, and LED applications.

Knowledge of the mechanical properties of glass is necessary for the majority of glass's technological applications. Synthesize (TWSm), (TWYb) glasses and measuring the elastic moduli of some (RE) doped tungsten-tellurite glasses are the motivation and novelty of the present work because elastic moduli (*L*), (*G*), (*K*), (*E*) are crucial parameter in manufacturing optical fibers. Moreover, ( $\sigma$ ), ( $\theta_D$ ), ( $T_S$ ), (*H*),



**Figure 1.** Photographs of the prepared samples: (a) for TWSm glass system and (b) for TWYb glass system.

(d), and (Z) will be calculated.

# 2. Experimental work

Two tellurite glass series have been synthesized (TWSm) and (TWYb). Oxides of high purity of tellurium dioxide, TeO<sub>2</sub> (Alfa Aesar 99.99), WO<sub>3</sub> (Alpha Chemika 99.9),  $Sm_2O_3$  (SIGMA-Aldrich 99.9), and  $Yb_2O_3$  (SIGMA-Aldrich 99.9) in specified weights were taken and grounded



20 (degree)

**Figure 2.** (a) XRD pattern of TWSm glass system and (b) of TWYb glass system.

using agate mortar for 15 minutes to get a homogeneous mixture. The batches were collected in a ceramic crucible which was preheated in the furnace up to 400°C for 30 min to reduce the tendency of vitalization. Stirring was done every 10 minutes to remove bubbles and ensure the homogeneity of the melt. The temperature was raised to 850°C depending on the glass composition. The glasses were cast rapidly into a pre-heated stainless mold (350°C) to remove thermal strains and then transferred to the annealing furnace at 290°C (1 hour). The annealed glass samples were polished for physical measurements. XRD Shimadzu diffractometer with Cu K $\alpha$  radiation at 0.154 nm has been used to confirm the amorphous state.

The glass density has been calculated at room temperature by the next equation:

$$\boldsymbol{\rho} = \left(\frac{W_a}{W_a - W_t}\right) \boldsymbol{\rho}_t \quad (\text{g.cm}^{-3}) \tag{1}$$

where  $\rho_t$  refers to the density of toluene,  $\rho =$  glassdensity,  $W_a$  is taken as the weight of the glass in air and  $W_t$  in toluene. Molar volumes ( $V_M$ ) have been calculated according to:

$$V_M = \frac{M}{\rho}$$
 (cm<sup>-3</sup>/mol) (2) and  $Z = \rho V_l$ .



**Figure 3.** (a) Density and molar volume of TWSm glasses and (b) of TWYb glasses.

The ultrasonic velocities were measured by using the pulseecho technique where the time intervals of two echoes were recorded by using Krautkramer USM 36 L model. The velocity was consequently obtained:

$$v = \frac{2d}{\Delta t} \qquad (\mathrm{cms}^{-1}) \tag{3}$$

where d = thickness of samples (cm), and  $\Delta t$  is the transit time between two successive echoes. Elastic moduli (EM);

$$\begin{split} (L) &= V_L^2 \rho, & (G) = V_S^2 \rho, \\ (K) &= L - \frac{4}{3}G, & (E) = 2G(1 + \sigma), \\ (\sigma) &= \frac{L - 2G}{2(L - G)}, & (\theta_D) = 251.2V_m \left(\frac{nM}{\rho}\right)^{\frac{1}{3}}, \\ (H) &= \frac{(1 - 2\sigma)E}{6(1 + \sigma)}, & (T_S) = \frac{V_S^2 M}{\psi^2 n}, \\ (V_m) &= \left[\frac{1}{3}\left(\frac{1}{V_l^3} + \frac{2}{V_S^3}\right)\right]^{-\frac{1}{3}}, & d = \frac{4G}{k}, \end{split}$$

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**Figure 4.** Longitudinal velocity  $(V_L)$  and shear velocity  $(V_S)$  of (a) TWSm glasses and (b) TWYb glass system.

# 3. Results and discussion

The produced glasses were clear, transparent, and have yellowish color for the two glass systems (TWSm) and (TWYb) as shown in Figure 1. While the X-ray diffraction profile of

**Table 1.** Density and molar volume of (TWSm), (TWYb)glasses.



**Figure 5.** Elastic modulus of (a) TWSm glass system and (b) of TWYb glass system.

the prepared glasses with no sharp peaks is in Figure 2. Both  $(\rho)$  and  $(V_M)$  of the (TWSm) and (TWYb) are represented in Figure 3. Values of  $(\rho)$  and  $(V_M)$  for Sm glass system are found in the range 6.00 g.cm<sup>-3</sup> to 6.12 g.cm<sup>-3</sup> and 28.98 (cm<sup>3</sup>.mol<sup>-1</sup>) to 29.39 (cm<sup>3</sup>.mol<sup>-1</sup>) respectively as shown in Table 1. Also, in the case of Yb glass system, these

**Table 2.** Longitudinal, shear and mean ultrasound velocitiesof (TWSm), (TWYb) glasses.

X Sm <sub>2</sub> O <sub>3</sub> (mol%)	( <b>p</b> )	$(V_M)$				
	$g.cm^{-3}$	$cm^3.mol^{-1}$	X Sm <sub>2</sub> O <sub>3</sub> (mol%)	$V_L$ (m/s)	$V_S$ (m/s)	$V_m$ (m/s)
0.0	6.0057	28.98	0.0	3309	1967	2177
1.0	6.0296	29.05	1.0	4031	2319	2575
2.0	6.0601	29.11	2.0	3844	2312	2557
3.0	6.0851	29.18	3.0	4089	2333	2592
4.0	6.0967	29.31	4.0	4100	2338	2598
5.0	6.1201	29.39	5.0	4337	2370	2642
X Yb <sub>2</sub> O <sub>3</sub> (mol%)			X Yb <sub>2</sub> O <sub>3</sub> (mol%)			
0.0	6.0057	28.98	0.0	3309	1967	2177
1.0	6.011	29.22	1.0	3969	2283	2535
2.0	6.0472	29.32	2.0	4371	2398	2673
3.0	6.07	29.48	3.0	3788	2244	2485
4.0	6.105	29.57	4.0	3375	1896	2109
5.0	6.138	29.68	5.0	4061	2332	2590

X Sm <sub>2</sub> O <sub>3</sub> (mol%)	EM (GPa) L	G	K	E	(σ)	( <i>H</i> ) (GPa)
0.0	65.76	23.24	34.78	57.01	0.227	4.23
1.0	97.97	32.43	54.74	81.24	0.253	5.35
2.0	89.54	32.39	46.35	78.82	0.217	6.12
3.0	101.74	33.12	57.58	83.38	0.258	5.33
4.0	102.49	33.33	58.05	83.92	0.259	5.35
5.0	115.12	34.38	69.28	88.49	0.287	4.88
X Yb <sub>2</sub> O <sub>3</sub> (mol%)						
0.0	65.76	23.24	34.78	57.01	0.227	4.23
1.0	94.69	31.33	52.92	78.49	0.253	5.16
2.0	115.53	34.77	69.17	89.35	0.285	4.99
3.0	87.01	30.57	46.34	75.17	0.229	5.51
4.0	69.54	21.95	40.28	55.72	0.269	3.37
5.0	101.23	33.38	56.72	83.72	0.254	5.47

**Table 3.** Experimental values of elastic moduli, Poison's ratio ( $\sigma$ ), Micro hardness (*H*), of (TWSm), (TWYb) glasses.

values are found in the range 6.00 g.cm<sup>-3</sup> to 6.13 g.cm<sup>-3</sup> and 28.98 (cm<sup>3</sup>.mol<sup>-1</sup>) to 29.68 (cm<sup>3</sup>.mol<sup>-1</sup>) respectively as shown in Table 1. For both systems, the values are found to increase as the amount of Sm<sub>2</sub>O<sub>3</sub> and Yb<sub>2</sub>O<sub>3</sub> is increased. Sm<sub>2</sub>O<sub>3</sub> and Yb<sub>2</sub>O<sub>3</sub> ions have high molecular weights than other oxide ions in the glass composition (TeO<sub>2</sub> = 159.6 g/mol, WO<sub>3</sub> = 231.84 g/mol, Sm<sub>2</sub>O<sub>3</sub> = 348.72 g/mol and Yb<sub>2</sub>O<sub>3</sub> = 394.08 g/mol). So, the replacement of WO<sub>3</sub> ions by Sm<sub>2</sub>O<sub>3</sub> or Yb<sub>2</sub>O<sub>3</sub> ions increases molecular weight and density. Depending on how the samarium ions or ytterbium ions settle themselves in the network, there are various explanations for this increase. The Sm<sup>3+</sup> and Yb<sup>3+</sup> ions may fill the network's interstitial spaces, which would result in structural compaction and an increase in density [20–36]. Longitudinal  $V_L$  and shear  $V_S$  ultrasonic velocities of TWSm glass samples are found in the range of 3309 - 4337 m/s and 1967 - 2370 m/s, respectively. While in the case of the TWYb glass system, the calculated ultrasonic longitudinal and shear velocities values are found in the range of 3309 - 4061 m/s. Table 2 and Figure 4 show the variation of the ultrasonic longitudinal and shear velocities with varying Sm<sub>2</sub>O<sub>3</sub> and Yb<sub>2</sub>O<sub>3</sub> mol%. As shown in Figure 4(a), for TWSm glass system, at y = 1 Mole fractions the two velocities exhibit an expected reduction from 4031 m/s to 3844 m/s and 2319 m/s to 2312 m/s, respectively. In the TWYb glass system shown in Figure 4(b), the two velocities showed an increasing trend with the Yb<sup>3+</sup> (2 mol%), then decreased at concentrations 3 and 4 mol% and begin to in-

**Table 4.** Softening temperature ( $T_S$ ), Debye temperature ( $\theta_D$ ), Fractal bond connectivity (d) and acoustic impedance (Z) of (TWSm), (TWYb) glasses.

X Sm <sub>2</sub> O <sub>3</sub> (mol%)	$T_{S}(\mathbf{k})$	$\theta_D(\mathbf{K})$	d	$Z \times 10^7  (\text{Kg m}^{-2} \text{ s}^{-1})$
0.0	817	262	2.67	1.99
1.0	1140	310	2.37	2.43
2.0	1137	308	2.76	2.32
3.0	1162	312	2.30	2.49
4.0	1171	313	2.29	2.50
5.0	1207	318	1.98	2.65
X Yb <sub>2</sub> O <sub>3</sub> (mol%)				
0.0	817	262	2.67	1.99
1.0	1107	304	2.37	2.39
2.0	1229	321	2.01	2.64
3.0	1083	298	2.64	2.29
4.0	778	253	2.18	2.06
5.0	1183	310	2.35	2.49



**Figure 6.** Poisson's ratio of (a) TWSm glasses and (b) of TWYb glasses.

crease again at 5 mol%. Both  $V_L$  and  $V_S$  could be compared with other glass series [20–35].

Table 3, Table 4, and Figures 5-9 show the variation of (EM) for both (TWSm) and (TWYb) glass series. For (TWSm) glass series; longitudinal modulus (L), 65.76 – 115.12 GPa, Young's modulus (E), 57.01 – 88.49 GPa, bulk modulus (K), 34.78 - 69.28 GPa, and shear modulus (G), 23.24 - 34.38 GPa. in the case of (TWSm) glass system. For (TWYb) glass series, the longitudinal modulus, L, 65.76 - 101.23 GPa, Young's modulus, E, 57.01 – 83.72 GPa, bulk modulus, *K*, 34.78 – 56.72 GPa and shear modulus, G, 23.24 - 33.38 GPa. With an increase in velocity or density, the (elastic modulus) EM increases. For the TWYb glass, EM showed an increasing trend with the Yb<sup>3+</sup> mol% until 2.0 mol%, then decreased and increased again at a concentration of 5 mol%. Therefore, for these samples, the network bond type has a vital role in determining the (EM) rather than the density. The present values of bulk modulus are in the range with other tellurite glasses;  $60B_2O_3$ .10Na<sub>2</sub>O.20TeO<sub>2</sub>.(10 - x) CaO $xZrO_2$ , [(TeO<sub>2</sub>)0.7(B<sub>2</sub>O<sub>3</sub>)0.3)0.7(ZnO)0.3]1 - x(SrO)x, SrO-B<sub>2</sub>O<sub>3</sub>-TeO<sub>2</sub>-V<sub>2</sub>O<sub>5</sub>-MnO<sub>2</sub>-MoO<sub>3</sub>-Tl<sub>2</sub>O<sub>3</sub>, B<sub>2</sub>O<sub>3</sub>-TeO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub>-ZnO-MgO-Li<sub>2</sub>O,  $xEu_2O_3$ -5PbO-25TeO<sub>2</sub>-(70 - x) B<sub>2</sub>O<sub>3</sub>, [(TeO<sub>2</sub>)0.7 (B<sub>2</sub>O<sub>3</sub>)0.3]1 - x [MnO<sub>2</sub>]x, xNb<sub>2</sub>O<sub>5</sub>-(1 -

x)TeO<sub>2</sub>, 12Bi<sub>2</sub>O<sub>3</sub>-8BaO-12ZnO-0.5CeO<sub>2</sub>-17.5SiO<sub>2</sub>-(50 – x) B<sub>2</sub>O<sub>3</sub>-xTeO<sub>2</sub> glasses with x = 0, 10, 20, 30 and 40 mol [27, 28, 30–32, 34–36] and far away from 75TeO<sub>2</sub>-5Na<sub>2</sub>O-(20 – x)TiO<sub>2</sub>-xBaO and GeO<sub>2</sub>-TeO<sub>2</sub>-Ga<sub>2</sub>O<sub>3</sub> [29, 33].

Values of  $(\sigma)$ ,  $(\theta_D)$ ,  $(T_S)$ , micro-hardness (H), f (d), and (Z) for both series. For TWSm samples,  $(\sigma)$ , increased from 0.227 at y = 0 M fraction to 0.287 at y = 5 mol%. A slight decrease is observed for ( $\sigma$ ), at y = 2 M fraction respectively. Similar behavior is observed in TWYb samples but here the decrease in ( $\sigma$ ), is observed at concentrations 3 and 5 mol%. Poisson's ratio depends on both the density of the cross-links and the dimension of the glass network. Also, showed (H) increased from 4.23 GPa to 6.12 GPa as  $Sm^{3+}$  increases from 0.0 to 2.0 mol% in the TWSm. Microhardness decreases from 5.33 to 4.88 GPa. Figure 7(b) showed the variation of the microhardness (H) in the TWYb glass system. It's observed that (H) has an increasing trend from 4.23 to 5.51 GPa. Meanwhile, the decrease is observed at y = 2 and 4 mol%, respectively. Besides that,  $(T_S)$ ,  $(\theta_D)$ , (d), and (Z) for both (TWSm) and (TWYb) glass series. In TWSm glass system,  $(T_S)$  and  $(\theta_D)$  temperatures increased with only a slight decrease observed at y = 2



**Figure 7.** Micro hardness (H) of (a) TWSm glasses and (b) of TWYb glasses.



**Figure 8.** Debye and softening temperatures of (a) TWSm glass system and (b) Debye and of TWYb glass system.

mole% with an increase in Sm<sup>3+</sup> content and their values varied from 262 – 318 (K) and 817 – 1207 (K) respectively. But for TWYb system, their values are found in the range 262 – 310 (K) and 817 – 1183 (K) respectively. The reduction of ( $\theta_D$ ) and ( $T_S$ ) temperatures depend on the number of atoms/unit volume and the presence of NBOs in the glass system [20–36].

The behavior of (d) and (Z) of the two glass series (TWSm) and (TWYb). It can be observed that for (TWSm) samples, parameter (d) decreased from 2.67 to 2.37 and 2.30 to 1.98 respectively with an increase in Sm<sup>3+</sup> content. Meanwhile, the acoustic impedance (Z) increased from  $1.99 \times 10^7$  to  $2.65 \times 10^7$  (Kg m<sup>-2</sup> s<sup>-1</sup>) with an increase in Sm<sup>3+</sup> content. For the other system (TWYb), parameter (d) exhibits a decreasing trend with only a slight increase observed with an increase in  $Yb^{3+}$  content. The acoustic impedance (Z) showed an increasing trend from  $1.99 \times 10^7$  to  $2.64 \times 10^7$ (Kg m<sup>-2</sup> s<sup>-1</sup>) with an increase in Yb<sup>3+</sup> content and then decreased to  $2.01 \times 10^7$  (Kg m<sup>-2</sup> s<sup>-1</sup>) at concentration 4 mol%. The decreasing values for fractal bond connectivity (d) can be linked to the increase in the amount of BOs and the decline in glass rigidity, respectively. Fractal bond connectivity (d) values only increase when y = 2.0 M fractions. The increase in (d) indicates an increase in the number of



**Figure 9.** (a) Fractal bond connectivity d, acoustic impedance Z for TWSm glasses and (b) for TWYb glasses.

NBOs [20–36]. Because of the high demand for communications, rise the development of potential glass materials for optical communications and the current data is crucial. The present outcome will complete the previous work on TeO<sub>2</sub>-WO<sub>3</sub> glass system with different modifiers such as GeO<sub>2</sub>, Ag<sub>2</sub>O and Bi<sub>2</sub>O<sub>3</sub> [37–39].

# 4. Conclusion

Because, mechanical properties are crucial parameter in manufacturing optical fibers, two glass series (TWSm) and (TWYb) were successfully synthesized. The outcome of this work can be summarized through the following points: 1. Both ( $\rho$ ) and ( $V_M$ ) increased with the increasing of dopants (Sm<sup>3+</sup>) or (Yb<sup>3+</sup>) concentration in both glass series,

2. Longitudinal  $V_L$ , and shear  $V_S$  velocities were examined and found in the range from 3309 to 4337 (ms<sup>-1</sup>), 3309 to 4061 (m<sup>-1</sup>), and 1967 to 2370 (m<sup>-1</sup>), 1967 to 2332 (m<sup>-1</sup>) for TWSm and TWYb, respectively,

3. The (*L*), (*G*), (*K*), and (*E*) show a similar trend with ultrasonic velocities with the increment of the  $\text{Sm}^{3+}$  or  $\text{Yb}^{3+}$  concentrations. Moreover, the bulk modulus of (TWSm) is higher than that of (TWYb) glass,

4. The ( $\sigma$ ) range from 0.227 to 0.287 for TWSm and from 0.227 to 0.287 for TWYb glasses,

5. The (H) range 4.23 GPa to 6.12 GPa for Sm samples and from 4.23 GPa to 5.51 GPa for Yb samples with the addition of dopants, which indicates the rigidity of the glass sample has been improved,

6. Both  $(\theta_D)$  and  $(T_S)$  temperatures were found in the range from 262 to 318 and 817 to 1207 K for (TWSm) and from 262 to 310 and 817 to 1183 K for (TWYb), respectively. The rising values of  $(\theta_D)$  and  $(T_S)$  temperatures can be predicted as a result of the strengthening of the glass structure which is due to the generation of BO,

7. The (*d*) showed a decreasing behavior with the addition of the  $\text{Sm}^{3+}$ .

### **Conflict of interest statement:**

The authors declare that they have no conflict of interest.

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